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281 CHAPTER14 SOLUTIONS MANUAL. 42. Explain on a particle basis why the vapor pressure of a solution is lower than a pure

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solvent. When a solvent contains a solute, fewer solvent particles occupy the surface.

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CHAPTER 14: CHEMICAL EQUILIBRIUM 453 We find K_c (for this system $\Delta n = +1$) $3.1 \times 10^{-9.3}$ (0.0821 298) $-\Delta \times = = = \times \times 4 c 3.8 \times 10^4 P P n K K RTRT K 14.28$ In this problem, you are asked to calculate K_c . Step 1: Calculate the initial concentration of NOCl.

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14.2 What are reducing sugars? Ans. Carbohydrates which reduces Fehling's solution to red precipitate of Cu_2O or Tollen's reagent to metallic Ag are called reducing sugars. All monosaccharides (both aldoses and ketoses) and disaccharides except sucrose are reducing sugars. Thus, D - (+) - glucose, D-(-)-fructose, D - (+) - maltose and D - (+) - lactose are reducing sugars.

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science which deals with the chemical changes in the environment. It includes our surroundings such as air, water, soil, forest etc.

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211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment
page 372 5. Compare the mass of the reactants and the mass of
the products in a chemical reaction, and explain how these
masses are related.

Chemistry Chapter 8 Solutions Manual

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