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12.3 Arrange the following compounds in increasing order of their boiling points:
CH₃CHO, CH₃CH₂OH, CH₃OCH₃,
CH₃CH₂CH₃ Ans. The order is : CH₃CH₂CH₃ < CH₃OCH₃ < CH₃CHO < CH₃CH₂OH All these compounds

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have comparable molecular masses CH₃CH₂OH undergoes extensive intermolecular H-bonding and thus its b.pt. is the highest. CH₃CHO is more polar than CH₃OCH₃ so ...

NCERT Solutions For Class 12 Chemistry Chapter 12 ...

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CHAPTER 12 REVIEW Solutions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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